

## **PRODUCT INFORMATION SHEET**

# Amplite® Colorimetric Zinc Ion Quantitation Kit

Catalog number: 19001 Unit size: 200 Tests

Component	Storage	Amount (Cat No. 19001)
Component A: Zn-620™	Freeze (< -15 °C), Minimize light exposure	1 vial (50 μL)
Component B: Assay Buffer	Freeze (< -15 °C)	1 bottle (20 mL)
Component C: ZnCl2 Standard	Freeze (< -15 °C), Minimize light exposure	1 vial (100 mM, 100 μL)

#### OVERVIEW

Zinc is an essential trace mineral element that plays an important role in a number of biological processes. It is an essential factor required for many enzymes, protein structures, and control of genetic expression. Zinc status also affects basic processes of cell division, growth, differentiation, development, and aging. Clinical signs of zinc deficiency include acrodermatitis, low immunity, diarrhea, poor healing, stunting, hypogonadism, fetal growth failure, teratology and abortion. Simple, direct and automation-ready procedures for measuring are highly desirable in research and drug discovery. AAT Bioquest's Amplite® Colorimetric Zinc Quantitation Kit provides a simple method for detecting zinc concentration in biological samples using our proprietary Zn-620<sup>™</sup>, in which Zinc binds to the probe with the enhanced absorption around 620 nm. The Zinc probe exhibits a large increase in 620 nm absorption in response to Zn2+ (>100 folds). Our kit formulation has enhanced Zn2+-specificity with little responses to other metals, e.g., Ca2+ and Mg2+. The assay can be used with biological samples such as serum, plasma, and urine with detection sensitivity at 1 µM. Our Amplite® Fluorimetric Zinc Ion Quantitation Kit (#19000) is even more sensitive, and can be used for detecting as low as 0.1 uM zinc ion.

## AT A GLANCE

#### Protocol Summary

- 1. Prepare  $Zn^{2+}$  Standards or test samples (50  $\mu$ L)
- 2. Add Zinc working solution (50 µL)
- 3. Incubate at room temperature for 5 10 minutes
- 4. Read absorbance ratio of A<sub>610nm</sub>/A<sub>470nm</sub>

#### Important Note

Thaw all kit components at room temperature before starting the experiment.

KEY PARAMETERS	
----------------	--

Absorbance microplate reader		
Absorbance	610/470 nm	
Recommended plate	Clear bottom	

## **PREPARATION OF STOCK SOLUTIONS**

Unless otherwise noted, all unused stock solutions should be divided into single-use aliquots and stored at -20 °C after preparation. Avoid repeated freeze-thaw cycles

#### ZnCl<sub>2</sub> standard solution (1 mM)

Add 10  $\mu L$  of 100 mM ZnCl\_2 Standard solution (Component C) into 990  $\mu L$  Assay Buffer (Component B) to get 1 mM ZnCl\_2 standard solution.

#### PREPARATION OF STANDARD SOLUTIONS

For convenience, use the Serial Dilution Planner: https://www.aatbio.com/tools/serial-dilution/19001

#### ZnCl<sub>2</sub> standard

Add 100  $\mu$ L of 1 mM ZnCl2 standard solution to 900  $\mu$ L Assay Buffer (Component B) to get 100  $\mu$ M ZnCl2 standard solution (Zn1). Take 100  $\mu$ M ZnCl2 standard solution (Zn1) and perform 1:2 serial dilutions in Assay Buffer (Component B) to get serially diluted ZnCl2 standards (Zn7 - Zn2).

#### PREPARATION OF WORKING SOLUTION

Add 25 µL of Zn-620<sup>™</sup> (Component A) into 5 mL Assay Buffer (Component B) to make Zn working solution.

## SAMPLE EXPERIMENTAL PROTOCOL

**Table 1.** Layout of  $ZnCl_2$  standards and test samples in a clear bottom 96-well microplate. Zn= Zinc Standards (Zn1 - Zn7, 100 to 1.56  $\mu$ M), BL=Blank Control, TS=Test Samples.

BL	BL	TS	TS
Zn1	Zn1		
Zn2	Zn2		
Zn3	Zn3		
Zn4	Zn4		
Zn5	Zn5		
Zn6	Zn6		
Zn7	Zn7		

Table 2. Reagent composition for each well.

Well	Volume	Reagent
Zn1 - Zn7	50 µL	Serial Dilutions (100 to 1.56 µM)
BL	50 µL	Assay Buffer
TS	50 µL	test sample

- 1. Dilute the test sample to 1.56 100  $\mu\text{M}$  range with Assay Buffer (Component B).
- 2. Prepare ZnCl<sub>2</sub> standards (Zn), blank controls (BL), and test samples (TS) according to the layout provided in Tables 1 and 2. For a 384-well plate, use 25  $\mu$ L of reagent per well instead of 50  $\mu$ L.
- 3. Add 50  $\mu$ L of Zn working solution to each well of ZnCl<sub>2</sub> standard, blank control, and test samples to make the total ZnCl<sub>2</sub> assay volume of 100  $\mu$ L/well. For a 384-well plate, add 25  $\mu$ L of Zn working solution into each well instead, for a total volume of 50

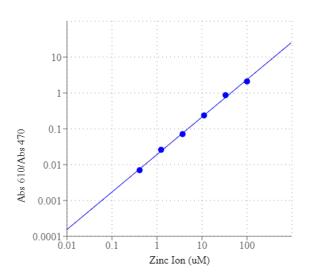
Tel: 408-733-1055 | Fax: 408-733-1304 | Email: support@aatbio.com

µL/well.

- 4. Incubate the reaction for 5 10 minutes at room temperature, protected from light.
- 5. Monitor the absorbance ratio increase with a absorbance plate reader at  $A_{610nm}/A_{470nm}$

## EXAMPLE DATA ANALYSIS AND FIGURES

Placeholder for image details



**Figure 1.** Zinc Chloride dose response was measured on a 96-well clear bottom plate with the Amplite® Colorimetric Zinc Ion Quantitation Kit.

#### DISCLAIMER

AAT Bioquest provides high-quality reagents and materials for research use only. For proper handling of potentially hazardous chemicals, please consult the Safety Data Sheet (SDS) provided for the product. Chemical analysis and/or reverse engineering of any kit or its components is strictly prohibited without written permission from AAT Bioquest. Please call 408-733-1055 or email info@aatbio.com if you have any questions.